

## pHun with Food – Lab Makeup

**Purpose:** Come up with a conclusion about what makes an acid an acid and a base a base according to their formulas.

**The following data was gathered during the pHun with food lab:**

### *Cabbage Juice Indicator*

*Purple - Neutral*

*Green - Basic*

*Pink - Acidic*

<i>Sample</i>	<i>Chemical Formula</i>	<i>Color Change</i>	<i>Acid or Base?</i>
<i>Water</i>	$H_2O$	<i>Purple</i>	
<i>Egg whites</i>	$Stiff-NH_2$	<i>Green</i>	
<i>Vinegar</i>	$HC_2H_3O_2$	<i>Pink</i>	
<i>Coca-cola</i>	$H_2CO_3$	<i>Pink</i>	
<i>Baking Soda</i>	$NaHCO_3$	<i>Green</i>	
<i>Salt</i>	$NaCl$	<i>Purple</i>	
<i>Milk of Magnesia</i>	$Mg(OH)_2$	<i>Green</i>	
<i>Lemon Juice</i>	$HC_6H_7O_7$	<i>Pink</i>	
<i>Peptobismol</i>	$C_7H_5BiO_4$	<i>Green</i>	
<i>Apple Juice</i>	$HC_4H_5O_5$	<i>Pink</i>	
<i>Tums</i>	$CaCO_3$	<i>Green</i>	
<i>Yogurt</i>	$HC_3H_5O_3$	<i>Pink</i>	

Poster:

Once you have come to a conclusion, make a poster to illustrate the differences between acids and bases. Things you might include are properties, examples, what makes an acid an acid, what makes a base a base.